Volumetric Methods of Analysis

Definitions and terms:

A <u>volumetric method</u> is one in which the analysis is completed by measuring the volume of a solution of known concentration needed to react completely with the substance being determined.

A <u>titration</u> is a process for determining the amount of a substance by measurement of the quantity of a reagent (the titrant) required to react completely with that substance.

A <u>standard solution</u> is a reagent of exactly known composition used in a titration.

A primary standard is a highly purified chemical compound.

Standardization:

Is a process where by the concentration of a standard solution is determined by titrating with a primary standard solution.

Requirements of a primary standard:

- 1. It must be of the highest purity.
- 2. It should be stable and not attacked by atmosphere.
- 3. It should not be hygroscopic.
- 4. It should be available and not too expensive.
- 5. It should have high equivalent to minimize weighing errors.

The <u>equivalence point</u>: is the point where the amount of standard solution added is chemically equivalent to the substance with which it reacts

The end point of a titration :-

Is the point at which physical changes associated with the equivalence point can be observed.

An <u>indicator</u>: is a chemical compound that exhibits a change in color as a result of concentration changes occurring near the equivalence point.

Concentration units:

Normality: Is the number of equivalent weights contained in one liter of solution.

<u>Molarity</u>: Is the number of gram molecular weights or the number of moles of solute in one liter of solution.

<u>Formality</u>: Is the number of gram formula weights per liter of solution.

Volumetric methods can be divided into four types:

- 1. Acid-base titrations.
- 2. Oxidation-reduction titrations.
- 3. Complexometric titrations.
- 4. Precipitation titrations.















COLD

DO NOT ENTER

STEP DOWN



DISTANCING





OVERHEAD CRANE







HIGH **TEMPERATURE**



CORROSION



AREA



MAGNETIC FIELD

AREA



HARD HAT **AREA**



UNDER CONSTRUCTION



HIGH **PRESSURE**



HOT SURFACE





Workplace Hazardous Materials Information System (WHMIS)

HAZARD SYMBOLS



Compressed Gas

Explosion Danger - Gas under pressure. May explode if heafed, punctured or dropped.



Health Hazard

May cause allergic reaction, cancer, birth defects, damage organs or harm fertility or unborn children.



Flammable Material

Potential Fire Hazard: Catches fire spontaneously if exposed to air or water or when exposed to heat sparks or flames or as a result of friction.



Harmful or Fatal

Acute Toxicity.

Potentially fatal poisonous substance if inhaled, swallowed, or through skin contact, even in small amounts.



Oxidizing

Fire and/or Explosion Risk in the presence of flammable or combustible material. May cause fire or enhance the combustion of other materials



Harmful

to skin, eyes or respiratory system. Fatal in large quantities. Hazardous to the Ozone Layer.



Explosion Hazard

Risk of explosion due to fire, shock, friction, heat or puncture



Harmful to the Environment

and/or aquatic life with long-lasting effects.



Corrosive

Causes severe Skin Burns & Eye Damage. Is corrosive to metal.



Biohazardous Infectious Materials

For organisms or toxins that can cause serious disease in people or animals resulting in illness or death.















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